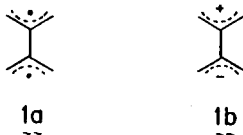


TANDEM ANNULATIONS.
2-BROMOMETHYL-3-(TRIMETHYLSILYLMETHYL)BUTA-1,3-DIENE
AS A SYNTHON FOR A ZWITTERION OF 2,2'-BIALLYL

Barry M. Trost and Roland Remuson
McElvain Laboratories of Organic Chemistry, Department of Chemistry
University of Wisconsin, 1101 University Avenue, Madison, WI 53706

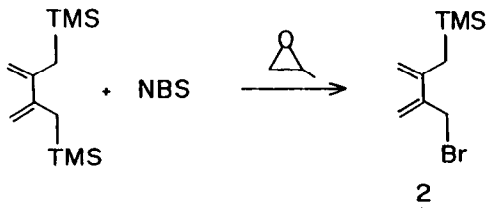
Summary: Alkylation of an enolate with the titled conjunctive reagent followed by cyclization of the allylsilane moiety onto the carbonyl group effects the first annulation; the resultant 2,3-dimethylene-cyclohexane participates in a subsequent Diels-Alder reaction to complete the second annulation.

Bifunctional conjunctive reagents that contain both nucleophilic and electrophilic centers whose reactivity is so adjusted to avoid self-annihilation offer the opportunity to create useful annulating agents.¹ Biallyl **1a**, a fascinating reactive intermediate,² offers the opportunity of

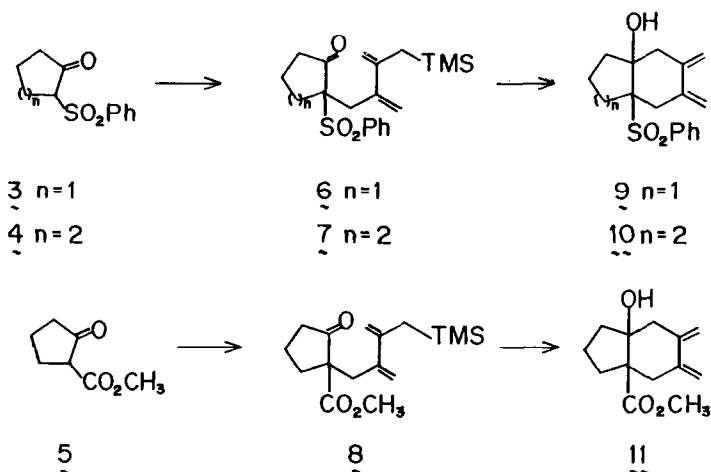


multiple ring annulations since an initial product would be a 2,3-disubstituted-1,3-butadiene, an excellent diene for a Diels-Alder reaction.^{3,4} We wish to report the synthesis of 2-bromomethyl-3-(trimethylsilylmethyl)-buta-1,3-diene (**2**), a synthon for a dipolar version of **1a**, *i.e.*, **1b**, and its use in a tandem annulation.

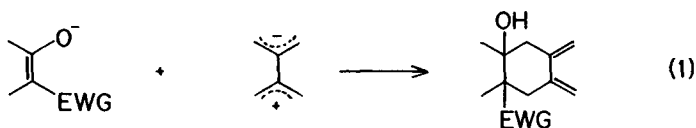
Subjection of 2,3-bis(trimethylsilylmethyl)-buta-1,3-diene⁴ to 1.15 eq of NBS in THF at -78° in the presence of propylene oxide gave the desired conjunctive reagent **2**⁵ in 40-50% yield after chromatography as a delicate oil



[IR 3110, 2970, 1595, 910, 885, 855. ^1NMR δ 0.0 (s, 9H), 1.8 (s, 2H), 4.2 (s, 2H), 5.0 (s, 1H), 5.2 (s, 1H), 5.3 (s, 1H), 5.42 (s, 1H)]. Alkylation of the ketosulfones **3** and **4** or β -ketoester **5** is best accomplished by treating their corresponding sodium or potassium salts (NaH or KH) with **2** in refluxing DME to give **6-8**⁵ in 50-60% yields. Characteristic absorptions in the NMR spectrum appear at δ 1.70 \pm 0.04 (s, C=CCH₂TMS), δ 2.64 \pm 0.11 and δ 2.92 \pm 0.11 (AB, J=15Hz, CH₂C=), δ 4.77 \pm 0.05, 4.88 \pm 0.02, 4.94 \pm 0.02, and 5.18 \pm 0.06 (each s, =C-H).

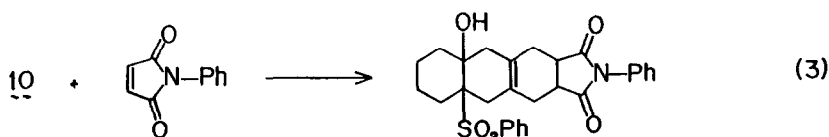
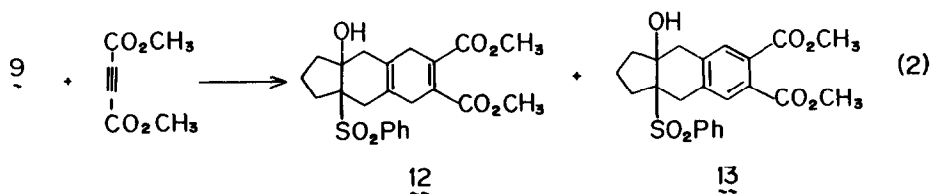


Cyclization to the 2,3-dimethylenecyclohexane occurs smoothly with either tetra-*n*-butylammonium fluoride (THF, 55 $^{\circ}$)⁶ or ethylaluminum dichloride (CH₂Cl₂, -78 $^{\circ}$)⁷ to give **9-11**⁵ in 60-78% yields, the latter reagent being preferred in the cyclization of **8** to **11**. Characteristic absorptions for the bis-methylene unit appear at δ 4.73 \pm 0.02, 4.80 \pm 0.05, 5.17 \pm 0.13, and 5.20 \pm 0.15. This two step sequence constitutes the equivalent of adding the diallyl unit to an electron rich olefin as symbolized by eq 1.

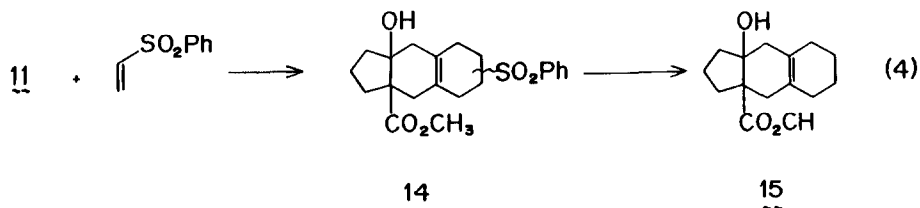


This formation of the dimethylenecyclohexane unit sets the stage for the second annulation, the Diels-Alder reaction. Dienes **9** and **10** cycloaddled with dimethyl acetylenedicarboxylate (eq 2, 95%) and *N*-phenylmaleimide (eq 3, 66%)

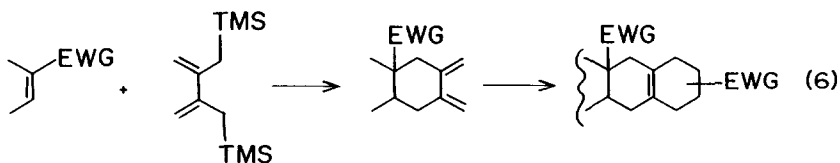
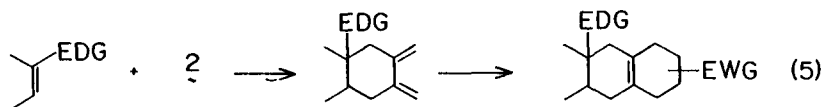
respectively in refluxing THF. In the former case, the initial adduct **12** was



contaminated by a small amount of its dehydrogenated product **13**. Diene **11** smoothly cycloaddled with phenylvinyl sulfone⁸ to give a regioisomeric mixture of adducts (eq 4, 80% yield). Chemoselective reductive desulfonation with buffered sodium amalgam⁹ gave the cyclopentano-1,2,3,4,5,6,7,8-octahydronaphthalene **15**⁵ in 71% yield.



The ability to generate multiple ring systems rapidly improves access to many of the complex molecular systems that are common synthetic targets. Tandem annulation offers one such approach. With the availability of the conjunctive reagent **2**, electron rich olefinic systems such as enolates (EDG - electron donating group) provide a ready access to bicycles according to eq 5.



This reagent offers an excellent complement to the recently reported sequence with 2,3-bis(trimethylsilylmethyl)buta-1,3-diene⁴ which allows such an elaboration beginning with the electronically reversed situation, *i.e.*, electron deficient olefins as in eq 6 (EWG - electron withdrawing group).

Acknowledgment. We thank the National Science Foundation for their generous support of our programs. We thank the CNRS for a fellowship to RR during his stay in our laboratories.

References

1. Cf (a) Trost, B.M.; Chan, D.M.T. J. Am. Chem. Soc. 1979, 101, 6429. (b) Trost, B.M.; Vincent, J.E. J. Am. Chem. Soc. 1980, 102, 5680. (c) Trost, B.M.; Curran, D.P. J. Am. Chem. Soc. 1981, 103, 7380. (d) Knapp, S.; O'Connor, V.; Mobilis, D. Tetrahedron Lett. 1980, 951.
2. Gajewski, J.J. J. Am. Chem. Soc. 1975, 97, 3457.
3. Cf Gaoni, Y.; Sadeh, S. J. Org. Chem. 1980, 45, 870 and earlier references cited therein. Butler, G.B.; Ottenbride, R.M. Tetrahedron Lett. 1967, 4873. Hamon, D.P.B.; Spurr, P.R. Synthesis, 1981, 873. Bellus, D.; von Bredaw, K.; Sauter, M.; Weiss, C.D. Helv. Chim. Acta, 1973, 56, 3004. Zahiereszhy, V.-I.; Musso, H. Annalen, 1973, 1777. Blomquist, A.T.; Verdol, J.A. J. Am. Chem. Soc. 1956, 78, 109; 1955, 77, 1806.
4. Trost, B.M.; Shimizu, M. J. Am. Chem. Soc. 1982, 104, 4299.
5. This compound has been characterized spectrally and its elemental composition determined by high resolution mass spectroscopy.
6. Sarkar, T.K.; Andersen, N.H. Tetrahedron Lett. 1978, 3513. Hosomi, A.; Shirahata, A.; Sakurai, J. ibid. 1978, 3043. Wetter, H. Helv. Chim. Acta, 1978, 61, 3072.
7. Trost, B.M.; Hiemstra, J. J. Am. Chem. Soc. 1982, 104, 886. For other Lewis acids see Deleris, G.; Dunoques, J.; Calas, R.; J. Organomet. Chem. 1975, 93, 43; Tetrahedron Lett. 1976, 2449. Hosomi, A.; Sakurai, H. ibid. 1976, 1295. Hosomi, A.; Endo, M.; Sakurai, H. Chem. Lett. 1976, 841. Ojima, I.; Miyazawa, Y.; Kumagai, M. J. C. S. Chem. Commun. 1976, 927. Itoh, K.; Fukui, M.; Kurachi, Y. ibid. 1977, 500. Tsunoda, T.; Suzuki, M.; Noyori, R. Tetrahedron Lett. 1980, 71.
8. Carr, R.V.C.; Paquette, L.A. J. Am. Chem. Soc. 1980, 102, 853.
9. Trost, B.M.; Arndt, H.C.; Strege, P.L.; Verhoeven, T.R. Tetrahedron Lett. 1976, 3477.

(Received in USA 12 December 1982)